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Why are the Raman Spectra of  
Crystalline and Amorphous Solids  
Different? Amorphous Materials:  
Structural Principles and

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Pharmaceuticals by DSC Analysis  
Mod-01 Lec-35 Amorphous and  
Crystalline State : Tg and Tm Easy  
tricks to learn difference between  
CRYSTALLINE and AMORPHOUS  
Solid. Difference between  
CRYSTALLINE and AMORPHOUS  
solid very easy AMORPHOUS AND  
CRYSTALLINE SOLIDS CLASS 12TH  
CHEMISTRY Solid State||  
Crystalline and Amorphous solid|  
Isotropic and Anisotropic nature  
of solids LECTURE-1 crystalline  
and amorphous solids in English

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12.1.1 Solid State: Amorphous  
and Crystalline Solids Is Glass a  
Liquid? Why is glass transparent?

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~~How to Understand Crystal  
Structures? The Structure of  
Crystalline Solids Chapter 3~~

~~Sulaiman May Ahmad~~

~~CLASSIFICATION OF CRYSTALLINE  
SOLIDS Determination of Crystal  
Structures Single Crystal,~~

Polycrystalline, Amorphous

{Texas A\0026M: Intro to

Materials} Crystalline Meaning

Material science concepts |

Crystalline Solids , Polycrystalline

solids , Amorphous solids Solid

State - Crystalline and Amorphous

Solids - English, Malayalam\_BASIC

CHEMISTRY. Lecture 04: X-ray

diffraction: Crystal structure

determination Characteristics of

crystalline and Amorphous solids

Doing Solids: Crash Course

# Get Free Characterization Of Amorphous And Crystalline #33 Crystalline and Amorphous Solids. BS 6th Principles And Applications Vol 37 Inorganic Material Chemistry CHEM-3115. By Dr. Asim Farid

XII-Chemistry-The Solid State-  
characteristics of solid,  
Amorphous and Crystalline Solids

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| SOLID STATE | TYPES OF SOLIDS  
| CRYSTALLINE AND AMORPHOUS  
SOLID | Crystalline Structure Part  
Three: Detecting Drug Excipient  
Incompatibility Characterization  
Of Amorphous And Crystalline  
Characterization of amorphous  
and crystalline ASR products  
formed in concrete aggregates 1.  
Introduction. Concrete damages  
due to alkali silica reaction (ASR)  
occur worldwide [ 1 ]. The  
expansion causing the... 2.  
Materials and methods. Concrete  
C1 and C2 were produced with a

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Characterization of amorphous  
and crystalline ASR products ...

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Amorphous and crystalline alkali  
silica reaction (ASR) products  
formed in aggregates of two  
different concrete mixtures  
exposed to the concrete prim test  
both at 38 °C and 60 °C have  
been analysed...

~~(PDF) Characterization of  
amorphous and crystalline ASR ...~~

The amorphous material shows a  
discharge profile typical of that of  
a single-phase material , whereas  
the crystalline tin shows the  
plateaus expected for a two-  
phase mixture. Both were cycled  
between voltage limits of and ,

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Vol 37

~~Characterization of Amorphous  
and Crystalline Tin-Cobalt ...~~

Optimization and characterization  
of amorphous/crystalline silicon  
heterojunction solar cells. N.

Jensen. Institute of Physical  
Electronics, University of  
Stuttgart, Pfaffenwaldring 47,  
D-70569 Stuttgart, Germany.

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~~Optimization and characterization  
of amorphous/crystalline ...~~

Abstract. The facile synthesis of  
 $\text{Al}_2\text{O}_3$  in the amorphous and  
corundum phase on both glass  
and quartz substrates is reported.  
The synthesis was carried out via



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aerosol assisted chemical vapour deposition using Al (acac) 3 and methanol. The films were analyzed using XRD, SEM, UV-vis spectroscopy and XPS. The coatings were highly crystalline (when annealed) with low carbon contamination levels and a relatively featureless morphology that gave rise to ultra high transparency in the UV ...

~~Synthesis and material characterization of amorphous and ...~~

With respect to the characterization of amorphous carbon and nanocrystalline carbon films, three kinds of TEM imaging techniques are usually used. 3.3.1. Electron diffraction (ED) When the atoms plane space

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satisfies Bragg's Law  $d = n \lambda / 2 \sin \theta$  and some other conditions, the electron diffraction pattern can be obtained. The simplest ...

~~Characterization of amorphous  
and nanocrystalline carbon ...~~

Crystalline polymers are polymers that have a well-organized

structure. Morphology:

Amorphous polymers are made out of atactic polymer chains.

Crystalline polymers are made out of syndiotactic and isotactic polymer chains. Attraction Forces:

Amorphous polymers have weak attraction forces between

polymer chains. Crystalline polymers have strong attraction forces between polymer chains.

~~Difference Between Amorphous~~

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Different amorphous solids don't show very distinctive diffraction patterns, as their elemental components aren't arranged in regular arrays. □ Melting Point. Crystalline solids have sharp melting points, that is, they change into liquids at definite temperatures. Amorphous solids, on the other hand, are thought to be liquids at all temperatures. This is because, on being heated, they do not abruptly change into liquids, but instead soften and start flowing in a semisolid form.

## ~~Crystalline Vs. Amorphous Solids What's the Difference ...~~

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Yiping Zhao and others published  
Characterization of Amorphous

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## ~~Characterization of Amorphous and Crystalline Rough ...~~

Nano-thermal characterization of amorphous and crystalline phases in chalcogenide thin films with scanning thermal microscopy J. L. Bosse,<sup>1</sup> M. Timofeeva,<sup>2</sup> P. D. Tovee,<sup>3</sup> B. J. Robinson,<sup>3</sup> B. D. Huey,<sup>1</sup> and O. V. Kolosov<sup>3,a)</sup>  
1Department of Materials Science & Engineering, University of Connecticut, Storrs, Connecticut 06269-3136, USA

## ~~Nano-thermal characterization of amorphous and crystalline ...~~

Ponja, SD; Parkin, IP; Carmalt, CJ;

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(2016) Synthesis and material characterization of amorphous and crystalline ( $\alpha$ -)  $\text{Al}_2\text{O}_3$  via aerosol assisted chemical vapour deposition. RSC Advances, 6 (105) pp. 102956-102960.

~~Synthesis and material  
characterization of amorphous  
and ...~~

In addition to a detailed description of the characteristics of random rough surfaces, Experimental Methods in the Physical Sciences, Volume 37, Characterization of Amorphous and Crystalline Rough Surface-Principles and Applications will focus on the basic principles of real and diffraction techniques for quantitative characterization of the rough surfaces. The book thus

# Get Free Characterization Of Amorphous And Crystalline Rough Surface Includes the latest development on the characterization and measurements of a wide variety of rough surfaces.

## ~~Characterization of Amorphous and Crystalline Rough ...~~

Amorphous and crystalline electrochromic WO<sub>3</sub> films exhibit quite different optical properties during coloration process. In the present work, amorphous and crystalline electrochromic WO<sub>3</sub> films prepared by a solution method were characterized using X-ray diffraction, scanning electron microscope, and transmission electron microscope techniques. A double-layer model with sharp interfaces was ...

## ~~Optical characterization of the~~

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To address this, high transparency NMs (%T 380-1100nm  $\square$ 94-99%) were synthesized by magnetron sputtering, and the effects of crystalline/amorphous (AlN/SiO<sub>2</sub>, AlN/Al<sub>2</sub>O<sub>3</sub>) and amorphous/amorphous (TiO<sub>2</sub>/SiO<sub>2</sub>) interfaces were characterized by spectrophotometry, transmission electron microscopy, and nanoindentation. We demonstrate that tuning layer configurations for improved transmittance resulted in substantial variations in microstructure and multifunctional film properties.

Synthesis and characterization of optically transparent ...

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Tadalafil (TD), a phosphodiesterase-5 (PDE-5) inhibitor with poor oral bioavailability. The aim of the study was to prepare and characterize three crystalline polymorphs of TD (II, III, and IV) and the tadalafil amorphous form (TD-AM). TD polymorphs and TD-AM were prepared and characterized by polar ...

~~Characterization and Stability of Amorphous Tadalafil and ...~~  
Amorphous metallic alloys generally exhibit higher corrosion resistance than their crystalline counterparts, which makes the study of amorphous metallic alloys based on nickel important. In this work, amorphous alloys of Ni<sub>62</sub>Nb<sub>38</sub>, Ni<sub>59.24</sub>Nb<sub>37.76</sub>B<sub>3.00</sub>



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Principles And Applications  
Vol 37**  
and Ni<sub>58.1</sub>Nb<sub>38.9</sub>B<sub>3.0</sub>  
compositions, with crystalline and  
amorphous structures, produced  
by arc melting and melt spinning  
techniques were studied ...

The structure of a growth or an etch front on a surface is not only a subject of great interest from the practical point of view but also is of fundamental scientific interest. Very often surfaces are created under non-equilibrium conditions such that the morphology is not always smooth. In addition to a detailed description of the characteristics of random rough surfaces, Experimental Methods in the Physical Sciences, Volume 37,

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Principles and Applications will focus on the basic principles of real and diffraction techniques for quantitative characterization of the rough surfaces. The book thus includes the latest development on the characterization and measurements of a wide variety of rough surfaces. The complementary nature of the real space and diffraction techniques is fully displayed. Key Features \*

- \* An accessible description of quantitative characterization of random rough surfaces and growth/etch fronts
- \* A detailed description of the principles, experimentation, and limitations of advanced real-space imaging techniques (such as atomic force

# Get Free Characterization Of Amorphous And Crystalline Rough Surface microscopy) and diffraction techniques (such as light scattering, X-ray diffraction, and electron diffraction) \*

Characterization of a variety of rough surfaces (e.g., self-affine, mounded, anisotropic, and two-level surfaces) accompanied by quantitative examples to illustrate the essence of the principles \* An insightful description of how rough surfaces are formed \* Presentation of the most recent examples of the applications of rough surfaces in various areas

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Amorphous-crystalline silicon (a-Si:H/c-Si) heterojunctions have recently drawn much attention owing to their low-temperature fabrication and high-efficiency photovoltaics. a-Si:H/c-Si heterojunctions were studied for the first time using the constant photocurrent method (CPM). The doping concentration in the p-type a-Si:H was varied. CPM derived absorption for energies greater than 1.4 eV is observed to increase with decreasing dopant concentration in the p-layer. This is attributed to a decrease in the density of defect states in the amorphous layer and the interface. A model is proposed wherein the amorphous layer and the interface constitute one

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absorbing layer while the crystalline substrate forms the other absorbing layer. A combined defect density in the amorphous layer and interface of  $2.8 \times 10^{18} \text{ cm}^{-3} \text{ eV}^{-1}$  at 0.4 eV from the valence band edge was measured for our best device. By comparing the combined defect density with that of a single amorphous layer the defect density at the interface is inferred to be  $5 \times 10^{12} \text{ cm}^{-2}$ .

A comprehensive approach to qualitative and quantitative characterization of crystalline and amorphous constituent phases of a largely heterogeneous Class F fly ash is presented. Traditionally, fly ash composition is expressed as bulk elemental oxide contents

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as determined by x-ray fluorescence spectroscopy. However, such analysis does not discern between relatively inert crystalline phases and highly reactive amorphous phases of similar elemental composition. Analytical techniques commonly used in the fields of chemistry and materials science have been modified and applied to effectively characterize the discrete crystalline and amorphous phases in a Class F fly ash. X-ray diffraction has been used to identify the crystalline phases present in the fly ash, and the Rietveld quantitative phase analysis method has been applied to determine the relative proportion of each of these phases. A synergistic method of

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Scanning electron microscopy, energy dispersive spectroscopy, and multispectral image analysis has been developed to identify and quantify the amorphous phases present in the fly ash. High-resolution spatial element maps are generated through intensive energy dispersive spectroscopy. The maps are statistically combined, phase boundaries are delineated, and relative phase fractions are determined through multispectral image analysis. Controlled leaching experiments have been performed to quantify the reactivity of the constituent phases in alkaline environments. Fly ash samples were subjected to various alkaline solutions for periods ranging from 1 day to 365

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The fly ash was physically separated from the solution, the crystalline phase fractions were determined through Rietveld quantitative x-ray diffraction, and the amorphous phase fractions were determined through multispectral image analysis. The phase distributions of the leached fly ash were compared to those of the raw, unleached ash. The composition of the leachate solution was also analyzed using inductively coupled plasma -- optical emission spectroscopy. The application of x-ray diffraction, Rietveld quantitative phase analysis, scanning electron microscopy, energy dispersive spectroscopy, multispectral image analysis, and inductively coupled plasma -- optical



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Emission spectroscopy techniques  
to the characterization of fly ash  
has facilitated a more  
comprehensive understanding of  
the widely-used engineering  
material.

Nuclear magnetic resonance  
(NMR) analysis of the recovered  
products from a series of  
controlled explosive shock-  
loading experiments on quartz  
powders was performed to  
investigate shock-induced  
amorphization processes.  
Silicon-29 NMR spectroscopy is an  
excellent probe of the local  
bonding environment of silicon in  
minerals and is capable of  
detecting and characterizing

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amorphous and disordered components. NMR spectra obtained for the recovered material exhibit a narrow resonance associated with the shocked crystalline material, and a broad component consistent with an amorphous phase despite the absence of evidence for glass from optical microscopy. The NMR measurements were performed over a range of recycle times from 1 to  $3 \times 10^5$  S. Results demonstrate that the magnetization in both the crystalline and amorphous material following power-law behavior as a function of recycle time. The amorphous component dominates the spectra for short NMR recycle times due to its shorter relaxation time relative to

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the crystalline material. Fractal analysis of the power-law relations suggests a fractal dimension of 2 for the amorphous phase and 3 for the crystalline phase.

Provides a summary of non-equilibrium glassy and amorphous structures and their macro- and microscopic thermal properties. The book contains a carefully selected works of fourteen internationally recognized scientists involving the advances of the physics and chemistry of the glassy and amorphous states.

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